# Optimization of high-performance field emission rare earth tungsten alloy cathodes\*

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The cathodes, as the electronic emission source of all kinds of electronic vacuum devices and spacecraft potential control system, its performance not only affects the overall efficiency of the equipment, but also limits the most important factors of the system long life and high reliability. In the field of space propulsion, the principle of electron emission from conventional cathodes mainly consists of thermal emission and field emission. Therefore, based on first-principles calculations using density functional theory, this study constructs atomic models of W cathode surfaces doped with different rare earth atoms. Using a  $(2 \times 2 \times 1)$  W (001) surface model, 1 ML of O atoms is absorbed on the top site of the surface, followed by doping rare earth atoms (La, Ce, Y) into the vacancy sites of the W-O lattice. The work functions of the system with rare earth atom coverages of 0.5 ML and 1 ML were calculated. Through liquid phase synthesis, plasma discharge sintering, and heat treatment, nano-scale second phase rare earth oxides(La<sub>2</sub>O<sub>3</sub>, CeO<sub>2</sub>, Y<sub>2</sub>O<sub>3</sub>, etc.)-tungsten cathodes were produced. Different ignition experiments were designed to simulate various operating conditions. The cascade arc plasma source was used for mass-loss and lifetime prediction tests on the cathode materials. After testing, Scanning Electron Microscopy and Energy Dispersive Spectrum microscopic characterizations of the cathode materials were conducted to analyze their composition, morphology, and elemental distribution. Optimization results reveal that the W-La, W-Ce, and W-Y cathodes prepared with this method exhibit excellent ablation resistance and plasma bombardment endurance at high temperatures. The nanoscale dispersion of the doped phases endows the cathode with superior electron emission properties, enhancing the overall efficiency of the system. Under plasma density of  $1.0 \times 10^{19} \,\mathrm{m}^{-3}$  and working temperature of 2000 K, the projected lifetime of rare earth tungsten alloy cathodes exceeds 2000 hours.

Keywords: Field emission cathode, Rare earth tungsten alloy, First-principles, calculations, Work function.

#### I. INTRODUCTION

As the primary and neutralizing electron source of vari3 ous electronic vacuum devices and spacecraft potential con4 trol systems, the performance of the cathodes not only affects
5 the overall efficiency of the system, but is also the most im6 portant factor limiting long service life and high reliability of
7 the system [1, 2]. In the 1960s, cathodes were first applied to
8 electric propulsion at the Lewis Research Centre and Hous9 ton Research Laboratory in the U.S.A. After decades of de10 velopment, cathode technology has been greatly developed,
11 including characterisation of cathode working plume modes,
12 sputtering corrosion properties of cathode components, emit13 ter working mechanisms, and validation of cathode working
14 life [3–7].

Compared with various conventional thermal cathodes, field emission cathodes have a series of advantages, such as fast startup, room temperature operation, no preheating delay, and high current density, which has been applied in many fields such as electron beam lithography, vacuum diodes and space propulsion systems, and other fields [8]. At present, in the field of vacuum devices, the commonly used cathode materials are Ba-W cathodes, lanthanum hexaboride(LaB<sub>6</sub>) and dodecalcium heptaaluminate(C12 A7) cathode materials, as well as new cathode materials developed on the basis of this,

25 and the advantages of the application of different emitter ma-26 terials are different [9-12]. The Ba-W cathode consists of a mixture of sintered porous tungsten impregnated with barium oxide, calcium oxide, and alumina, where porous tungsten is the substrate and the mixture is the raw material. the 30 Ba-W cathode has an escape power of about 2.1 eV and an 31 operating temperature of about 1300 K, but it has high requirements for the working medium. the LaB<sub>6</sub> is generally 33 formed by the mechanical processing of powder pressurised 34 and sintered polycrystalline structured rods, and has an es-35 cape power of about 2.7 eV, the working temperature ex-36 ceeds 1800 K, and the long time high temperature service 37 makes its power consumption larger. In recent years, theo-38 ries have shown that C12 A7 compounds have lower escape  $_{99}$  power (theoretically  $0.6\,\mathrm{eV}$ ) and operating temperature (theoretically 900 K), and although no conclusion has been drawn from experiments, their excellent performance has attracted a great deal of attention from domestic and foreign researchers. Future cathode research will also focus on the electron es-44 cape work of the cathode emitter and the cathode operating 45 temperature [13, 14]. However, in the Magneto Plasma Dy-<sup>46</sup> namic Thruster(MPDT), the cathode, as a core component, is 47 in the center of the high-energy plasma plume, with an instan-48 taneous ignition voltage of up to several thousand volts, an 49 operating temperature higher than 1500K, and an extremely 50 harsh working environment, which puts forward high require-51 ments for the cathode material and structure [15].

Metallic materials are the earliest researched and most widely used field emission materials, mainly W, Mo, Ta, etc. However, pure W has high electron escape power and is prone to difficult arc initiation and breakdown at low voltage. However, pure W has a high electron escape power, which makes

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57 it easy to appear phenomena such as difficult to start an arc 109 58 and difficult to break down at low voltage. In addition, due 110 59 to La, Ce, Y and other rare earth elements have lower es-60 cape power and higher melting point, it has been shown that the addition of rare earth oxides with lower escape power to can significantly improve the ablation resistance of the cathode. In recent years, North American researchers and 64 scholars have prepared E3 ternary electrodes with ZrO<sub>2</sub> as 65 an additive, and the main components are W-La<sub>2</sub>O<sub>3</sub>-Y<sub>2</sub>O<sub>3</sub>- $_{66}$  ZrO $_{2}$ . Zhu Wenguang et al. compared the arc initiation per-  $_{115}$ 67 formance of E3 electrodes with that of traditional multiele-68 ment electrodes, and in the ablation at a current of 200 A for 69 5 h, E3 electrodes showed the smallest amount of loss by ab-70 lation, and the dimensional stability was good, and the degree 71 of recrystallisation was weak. Therefore, based on the tradi-72 tional cathode electron emission theory and traditional cathode material types this study will use the high melting point component W as the cathode substrate, doped with different 75 types and contents of rare earth elements, and complete its 76 various performance tests and ground prototype experiments 77 to verify [16–18].

Atomic models were constructed with tungsten(W)-O sur-79 faces doped with various rare earth atoms (La, Ce, Y), us-80 ing first-principles calculations and density functional the-81 ory(DFT). Calculations for the work functions were con-82 ducted for the models of 0.5 ML and 1.0 ML doping levels. 83 The results showed that doping rare earth elements greatly 84 lowered the work function of the alloy cathode, improving  $_{85}$  electron emission performance, and that  $0.5\,\mathrm{ML}$  doping in W-O lattice sites resulted in the lowest work function.

In the present study, nano-doped rare earth tungsten cathode materials were prepared using liquid-phase synthesis and plasma discharge sintering techniques. A series of ignition 90 tests on the thruster prototypes were conducted along with 134 microstructural characterization experiments. Electron emis-92 sion performance, ignition performance, and efficiency of a 136 rameters such as the material's work function, temperature, 93 tungsten alloy cathode doped with various elements and pro-94 portions were tested.

conditions, the long-life test of the whole machine is subject 143 the cathode. At lower temperatures, thermionically emitted to greater constraints. Therefore, this study independently 144 electrons that overcome the barrier contribute negligibly to conducts a life assessment experiment for cathode, and con- 145 the emission current, with the emission primarily consisting ducts a mass-loss-life prediction of cathode through a self- 146 of field-emitted electrons near the Fermi level [21]. developed cascade arc plasma generator source. The results 147 the cathode, the preliminary prediction of the rare-earth tung- 153 face and attempt to escape, they are hindered by this surface 108 sten alloy cathode life reaches 2000 h.

# ELECTRON EMISSION PRINCIPLE OF FIELD EMISSION CATHODE

The emission of cathodes used in thruster systems primar-112 ily relies on the principles of thermionic emission and field emission [19]. Thermionic emission follows the Richardson 114 equation:

$$j_0 = [AT_K^2] exp(-\phi_k/kT_K) \tag{1}$$

As shown in Equation (1),  $j_0$  represents the zero-field emission current density. A is the theoretical value of the material's emission constant,  $T_K$  is the cathode operating temperature, and  $\phi_k$  represents the material's work function [20].

Similar to the theoretical derivation of the thermionic emis-121 sion equation, Fowler and Nordheim developed the field 122 emission theory for metals. They assumed the following: 123 (1) the distribution of band electrons conforms to the Fermi-124 Dirac distribution; (2) a smooth, planar metal surface is considered, ignoring atomic-scale irregularities; (3) classical image forces affecting electrons are taken into account; (4) the 127 work function distribution is uniform. Under these assump-128 tions, the following equation holds:

$$j_0 = \frac{1.54 \times 10^{-6} \xi^2}{\phi_k} \exp \left[ -\frac{6.83 \times 10^7 \phi_k^{3/2}}{\xi} \theta(y_0) \right]$$
 (2)

In Equation(2),  $\xi$  represents the electric field strength, measured in V/cm. Where  $\theta(y_0)$  is a slow-variable function 132 of  $\xi$ ,

$$y_0 = \left(3.79 \times 10^{-4} \frac{\sqrt{\xi}}{\phi_k}\right) \tag{3}$$

According to the electron emission equations above, the 135 zero-field emission current density j<sub>0</sub> is closely related to paand electric field strength. Theoretically, the lower the work function of the cathode material, the more readily electrons within the material can overcome the surface potential barrier 140 to emit from the cathode surface. Additionally, the operat-In addition, based on the working principle of cathode for 141 ing temperature  $T_K$  and electric field strength  $\xi$  are directly various vacuum devices, in view of the existing experimental 142 proportional to the zero-field emission current density jo of

For the work function of cathode materials, the periodic arshow that the nano rare-earth tungsten alloy cathode has bet- 148 rangement of lattice ions is interrupted at the metal-vacuum ter electron emission performance than the conventional cath- 149 boundary, thereby disrupting the periodicity of the potential ode, in which the lanthanum oxide doped tungsten alloy with 150 field [22]. The potential energy increases in a specific manner different mass fractions makes the cathode material escape 151 and approaches zero at infinity, forming the surface potential work reduced significantly; after the independent life test of 152 barrier of the metal. When electrons move to the metal surbarrier, which is defined as the material's work function.

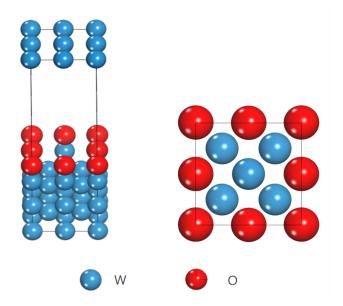


Fig. 1. Top-Site Adsorption of O Atoms on W(001) Surface.

$$\phi_{\mathbf{k}} = E_{\mathbf{V}} - E_{\mathbf{F}} \tag{4}$$

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All electrons attempting to escape from the metal must 156 157 have energy at least equal to the Fermi level plus the value of 158 the work function and follow a statistical distribution. Their average energy equals 32 KT, with each degree of freedom 159 contributing an average energy of 12 KT, consistent with the 160 results of kinetic molecular theory [23]. 161

On the other hand, the cathode evaporation rate increases 162 163 sharply with rising cathode operating temperature. Cathode evaporation directly impacts the cathode's lifespan, grid emission, and inter-electrode insulation performance. Ideally, a cathode should have high emission capability, requiring a low work function and minimal evaporation. Considering these two requirements, a quality factor F can be used to rep-169 resent the performance:

$$F = \phi_{\mathbf{k}} \times 10^3 / T_{\mathbf{e}} (eV/K)$$
 (5)

Te is the temperature (K) at which the material's vapor 172 pressure reaches  $10^{-5}$  mmHg. To ensure thruster performance, the cathode temperature should not exceed its "vapor pressure temperature" Te [24]. 174

Therefore, the selection and optimization of cathode emit-176 ter materials need to balance electronic emission performance 177 with thermodynamic properties. Higher electron emission 210 178 performance can enhance cathode discharge efficiency and 211 overall thruster efficiency, while better thermodynamic prop-180 erties extend the service life of the cathode under extreme op-181 erating conditions. The research in this paper will mainly start 213 from the simulation and experimental optimisation of cathode 214 physics and one of the greatest discoveries of the 20 th cenmaterial components to reduce the emitter escape work, and 215 tury. Using quantum mechanics, it is possible to explain and

185 emission performance, lifetime, and ablation performance, so as to further complete the research on the cathode for high-(4) <sub>187</sub> performance field-emission space propulsion.

On the basis of traditional field emission cathode research, 189 new field emission cathode has become a current research 190 hotspot. Researchers at the National University of Defense 191 Technology prepared a carbon fiber composite graphite cath-192 ode and tested its electron emission performance, which 193 showed that the field emission threshold electric field of the 194 40% (mass fraction) carbon fiber composite graphite cathode was reduced from 143 kV/cm to 119 kV/cm, a reduction of about 16.8%, and due to the characteristics of the structural 197 stability of the carbon fibers in the process of electron emis-198 sion, the carbon fibers conformity is also favours the improvement of cathode service life [25]. Maksim A Chumak et al, 200 Ioffe Research Institute, St. Petersburg, Russia, used atomic 201 layer deposition (ALD) to produce field-emission cathodes 202 with carbon nanotube (CNT) arrays coated with ultrathin 203 nickel oxide (CNT/NiO), and proposed for the first time to 204 reduce the figure of merit of the field-emission nanocompos-205 ite CNT/NiO cathode by changing the chemical composition 206 of the oxide coatings, and it was demonstrated that, according 207 to the secondary electron cut-off energy, the work function of 208 pure CNTs is 4.95 eV, and the work function of NiO layer 209 deposited on CNTs after heat treatment is reduced [26].

### III. FIRST-PRINCIPLES STUDY ON THE SURFACE WORK FUNCTION OF RARE EARTH TUNGSTEN ALLOY **CATHODE**

Quantum mechanics is an important foundation of modern verify the key parameters of cathode material such as electron 216 predict the physicochemical properties of a wide range of sys-

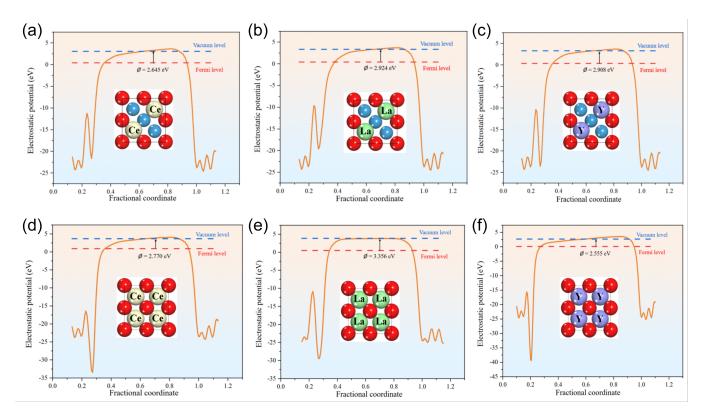


Fig. 2. W-O Crystal Surface Doped with rare earth atoms model and work function calculation.(a) Doped with 0.5 ML Ce; (b) Doped with 0.5 ML La; (c) Doped with 0.5 ML ML Y; (d) Doped with 1.0 ML Ce; (e) Doped with 1.0 ML La; (f) Doped with 1.0 ML Y.

217 tems and to quantitatively analyze the laws of their electronic 245 order to assess the possible effect the doped elements have on motion. The first principle is a computational method based 246 the current density of the cathode emission [30]. on quantum mechanics to study the properties of materials from the point of view of electron motion. The wave function contains all the information of the computational system, which greatly limits the scope of its practical applications, and the establishment of the density functional theory solves the problem of the complexity of the wave function. The basic idea of density functional theory is to change the characteristics based on the orbital wave function, with the particle density function to express the system base state of each physical quantity, to the electron density function represents the system energy [27, 28]. 229

Material Studio material simulation software incorporates 258 variety of three-dimensional scale simulation calculation methods, which can complete the cross-scale scientific research from the microscopic electronic structure to the macroscopic performance prediction [29]. It is on the basis of this advantage that the atomic-scale emission structure was modeled with the use of the Materials Studio in the current work. Geometric optimization was made for tungsten alloy models 242 tion, while computations of work function were performed in 270 cell are about 9.11 eV, 7.40 eV, and 8.20 eV, respectively; 243 LDA functional and PBE-GGA functional. The work func- 271 this means the O atoms are preferentially adsorbed on the top 244 tion of the (001) crystal plane for tungsten is calculated in 272 site shown in Figure 1 [32].

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The models mainly include the following: the adsorption 248 of O atoms and La atoms on a tungsten surface, adsorption of 249 O atoms and Ce atoms on a tungsten surface, and adsorption 250 of O atoms and Y atoms on a tungsten surface.

The current work has used the CASTEP density functional 252 calculation module of Materials Studio, which is based on a 253 plane-wave basis set. Among the well-known classical algorithms in CASTEP, the main one is the plane-wave pseudopotential method. By moving the model of a tungsten atomic structure and geometric optimization, it was possi-257 ble to find the ground state with the lowest energy. We used the default number of maximum steps, while the cutoff energy was 278.0 eV. The method of calculation was "Fine" and for the rest of the parameters that were given, we kept them the same as the default. Under the ultrasoft pseudopotential, GGA was being applied, and the functional from Perdew-Burke-Ernzerhof was picked to describe the electron exchange-correlated interactions [31].

O atomic layer was adsorbed on the surface of the supercell doped with various rare earth elements. Further on, the re-  $^{266}$  W(2  $\times$  2  $\times$  1). According to the related computational literalaxation of surface atoms was conducted and work function 267 ture, the larger the adsorption energy is, the more stable the was calculated under convergence conditions. The pseudopo- 268 adsorption system is. In fact, the adsorption energies of O tential method was realized for solution of Schrödinger equa- 269 atoms on the top site, bridge site, and hollow site of W super274 ing coverages were adsorbed. Similar to O atoms, rare earth 318 slightly reduced compared to 0.5 ML Y doping. 275 atoms on the  $(2 \times 2 \times 1)$  W(001)-O (top-site) surface also 276 have three possible adsorption positions. Due to the larger 277 atomic radius of rare earth elements, adsorption at the top and 319 bridge sites causes significant lattice distortion in the W lat- 320 tice. Computational results indicate that rare earth atoms are more likely to adsorb at hollow sites on the W-O (top-site) surface. The formula for calculating the adsorption energy of 322 282 rare earth atoms is as follows:

$$E_{ad} = -\frac{1}{N} (E_{La+W(001)-O(top)} - NE_{La} - E_{W(001)-O(top)})$$
(6)

As shown in Figure 2a to 2c, the work function for rare  $_{285}$  earth atoms in the W-O crystal with a coverage of  $0.5\,\mathrm{ML}$ was calculated using Equation (6). Because rare earth atoms readily lose their two outermost valence electrons, transferring them to the inner O atoms, the electron density of the 289 coverage layer is lower than that of the substrate surface layer. This results in a dipole layer with a positive charge on the outside, raising the surface potential and reducing the barrier height.

Similarly, this study further calculated the work function of W-O crystals with a rare earth atom coverage of 1.0 ML, 337 rare-earth tungsten alloy cathode raw material mixed pow-295 meaning that rare earth atoms fully occupy the hollow sites in 296 the W crystal, as shown in Figure 2d to 2f.

Table 1. Work Function of W-O (Top-Site) Doped with Different Rare Earth Atoms

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W(001)-O Top-Site Doping	Work Function (eV)
0.5 ML Ce atoms	2.645
0.5 ML La atoms	2.924
0.5 ML Y atoms	2.908
1.0 ML Ce atoms	2.770
1.0 ML La atoms	3.356
1.0 ML Y atoms	2.555

reduction of a dipole moment increases the work function.

cases, validating the feasibility of this study's approach to 363 process [34–36]. optimize cathode electron emission performance by doping 364 316 Y atom doping into the hollow sites of the tungsten crystal 370 experiments can be completed.

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On the W-O (top-site) surface, rare earth atoms with vary- 317 results in minimal lattice distortion, and its work function is

#### EXPERIMENTAL STUDY ON CATHODE OF RARE EARTH TUNGSTEN ALLOY

#### Synthesis and processing of rare earth tungsten alloy cathodes

For various vacuum electronics and space propulsion sys-324 tems, the primary requirements for cathodes are superior elec-325 tron emission capability and high ablation resistance to with-326 stand impacts from high-energy particles. Experimental research indicates that the electron emission performance, melt-328 ing point, and ablation resistance of rare earth tungsten alloy 329 cathodes are closely related to the chemical properties, phys-330 ical dispersion, and percentage content of the nano-doped 331 phase. We synthesized W-La<sub>2</sub>O<sub>3</sub> alloy cathodes with differ-332 ent doping levels and studied their electron emission performance at various temperatures and voltages for different W-<sup>334</sup> La<sub>2</sub>O<sub>3</sub> doping concentrations [33].

Figure 3 a shows the flow process for the preparation of 336 nano rare earth tungsten alloy cathode. In this study, the nano 338 ders were produced by two methods, namely, liquid-phase 339 synthesis reduction method and ball milling method, and 340 moulded with reference to the processing method of LaB<sub>6</sub> 341 cathode emitter. The liquid-phase synthesis method is to mix tungstate, lanthanate and complexing reagent in deionised water according to the measured ratios, and then react for about 8 hours at the appropriate temperature, and then spray-345 drying is carried out to obtain the homogeneous raw mate-346 rial powders; the ball milling method is to use the planetary ball mill to mill the mixed powders for 4 hours at 240 rpm under argon gas at room temperature, with a ball-to-powder weight ratio of 8:1. After mixing the raw material powders, 350 the powders were pressed into rods with a diameter of 16 mm As marked in Table 1, the calculational results of work 351 by cold isostatic pressing at a pressure of 150 MPa, and then function for W-O (top-site) doped with 0.5 ML and 1.0 ML 352 sintered at 2600 K for 4 h in a dry hydrogen atmosphere. Fiof different rare earth atoms show that the large atomic radius 353 nally, in order to remove the oxide layer from the samples of rare earth elements results in large lattice distortion when 354 and to obtain a smooth surface, we ground the rods to a difully occupying the hollow sites in the W crystal, and there- 355 ameter of 9.0 mm using a computer numerical control (CNC) fore such a system is unstable. When the number of adsorbed 356 machine. The densities of the W-La samples were obtained atoms increases above each optimal coverage, the interactions 357 using the Archimedes method, and the theoretical densities between dipoles increase gradually. Here, the middle atoms 358 were calculated based on the fraction of each component as are depolarized by an electric field of adjacent dipoles. The 359 a function of the density, and the densities of the W-La cath-360 odes obtained by the preparations in the present study were Comparing the data in the table, doping rare earth atoms 361 in the range of 98.5% - 99.5%. Figure 3b is the physical diinto the W crystal reduces the surface work function in all 362 agram of the W-La cathode prepared according to the above

After the cathode was processed, we measured its actual tungsten with rare earth elements, thereby enhancing the ef- 365 density using the Archimedes drainage method and compared ficiency of magnetoplasma thrusters. Additionally, the calcu-313 lation results show that doping 0.5 ML La or Ce atoms at the 367 rameters, as shown in Table 2 The densities higher than 98% 314 W-O (top-site) achieves the greatest reduction in work func- 368 proved that the sintered state of the cathode obtained by this 315 tion. Due to the relatively smaller radius of Y atoms, full 369 method is dense and uniform, and the subsequent tests and

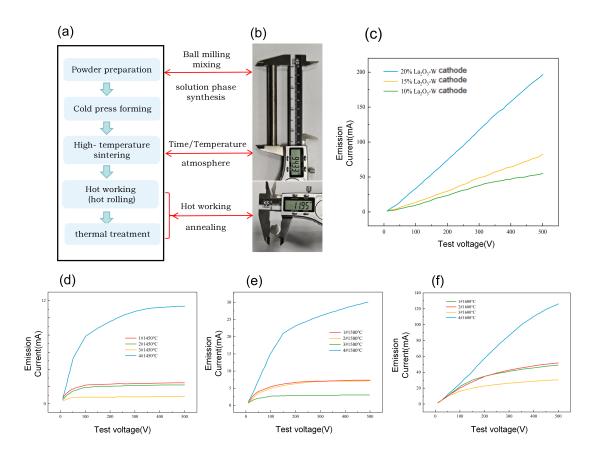


Fig. 3. (a)Processing workflow of rare earth tungsten alloy cathodes.(b)Physical drawing of cathode.(c)Electron emission properties of lanthanum oxide-doped tungsten cathodes with different contents.(d)-(f)Electron emission properties of different types of rare-earth cathodes and tungsten cathodes at 1450 °C,1500 °C and 1600 °C, respectively.1#:W-Zr-Y cathode;2#:W-Y cathode;3#:pure W cathode;4#:W-La15 cathode

The series of rare earth tungsten alloy cathodes we de- 393 to its lower electron escape work. 372 signed and processed are aimed at the extreme working en-373 vironment of vacuum equipments and space thrusters. By 394 374 optimizing the emitter composition, we aim to balance high 395 alloy cathodes with different doping ratios using the same 375 electron emission performance with ablation resistance for 396 processing method and made preliminary predictions of their 276 extended service life. Figure 3c shows electron emission per- 397 properties [37, 38]. During the sintering process of W-Ce cath-377 formance data for W cathodes doped with various mass frac- 398 ode, due to the volatilisation of CeO2, which affects the stanificantly, confirming that adding low work function compo- 401 sintering [39]. nents to optimize emitter performance is feasible. In addition, we tested the electron emission properties of different doping 382 types of cathodes at 1450 °C,1500 °C and 1600 °C, respec-383 tively, as shown in Figures 3d to 3f. The results show that the 384 emission current densities of the four cathodes tested in the 385 experiments increase to different degrees with the increase of 386 the test voltage and the test temperature, among which the W-387 La15 cathode has the largest increase; the pure W cathode no longer increases in the emission current density when the test voltage reaches the threshold value due to its lower electron 391 emission current density of the pure W cathode no longer in-392 creases when the test voltage reaches the threshold value due

In addition, we prepared W-Ce and W-Y rare earth tungsten tions of W-La<sub>2</sub>O<sub>3</sub>. As W-La<sub>2</sub>O<sub>3</sub> content increases, the overall 399 bility of the hollow cathode discharge, the process needs to electron emission performance of the cathode improves sig- 400 be strictly controlled, and it needs to be prepared by insulated

Table 2. Actual density and homogeneity of different types of cathodes.

Type of cathode	D <sub>A</sub> (g/cm <sup>3</sup> )	D <sub>T</sub> (g/cm <sup>3</sup> )	Homogeneity (%)
W-Y10 cathode	18.69	18.81	99.4±0.1
W-La10 cathode	18.87	18.98	$99.4 \pm 0.1$
W-La15 cathode	18.65	18.79	$99.3 \pm 0.1$
W-La20 cathode	18.46	18.62	$99.1 \pm 0.1$

## Independent service life experiment using cascade arc plasma source

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Under conditions of existing experimental possibilities, the 404 405 verification of the service life of cathodes is a long-time and validation conducted together with thrusters. As space missions have imposed longer lives on electric propulsion systems, full-life ground testing has become increasingly impractical. For instance, the ground-tested ion thruster for 466 NASA's \*Deep Space 1\* mission lasted for 30 352 hours, 467 which is more than five years [40]. In comparison, the JIMO mission would have used six ion thrusters powered by nuclear 468 energy as its main propulsion; these would have to individually be rated for 83,000-hour lives. Assuming an efficient test 470 factors affecting the lifespan and efficiency of the cathode maduration of 75 percent, a 1.5-times redundancy life test would take approximately 19 years [41]. The extended life test of the U.S. NEXT ion thruster for the subsequent deep space exploration mission started in 2005, and a total of 51 184 h life tests were conducted until 2017, consuming 918 kg of xenon gas. Therefore, under the existing experimental conditions, we will carry out a rapid life test method to evaluate the cathode life in combination with relevant experimental parame-423 ters, which also greatly reduces the life test time and cost [42]. 424

We have developed a cascade arc plasma source to simu-426 late the real working environment of rare earth tungsten alloy 427 cathodes. This consists of a vacuum system, the power sup-<sup>428</sup> ply system, the superconducting magnet system, the plasma <sup>483</sup> in the rare earth tungsten alloy cathode prepared by our ball 429 generation device, the gas supply system, the water cooling 484 milling method and liquid phase synthesis method are uni-430 system, and the Langmuir probe system, as shown in Figure 431 4a

In this paper, it is considered that the tip morphology of 433 the cathode changes significantly when the mass loss of the cathode reaches 10% - 15%. This reduces the effectiveness of 435 small-hole current limitation and leads to inability to maintain stable discharge when the propellant flow rate exceeds the set operating range, and the cumulative operating time under rated conditions cannot be achieved. Beyond this point, 493 pellant flow rate. This result suggests that, in order to obtain 439 the cathode is to be considered functionally degraded and at 494 the same or greater propulsive efficiency, the ignition voltage the end of its service life. 440

cascade arc plasma source to simulate the thruster's experi- 497 to the cathode ignition instant. As can be seen in the figure, mental environment by placing the cathode within the plasma 498 the cathode surface is left with inhomogeneous holes from the source and setting specific plasma density and temperature 499 depletion of La<sub>2</sub>O<sub>3</sub> particles, which indicates the feasibility parameters. Material ablation and service life predictions 500 of this study to improve the electron emission performance of were conducted based on mass loss over a specified experi-447 mental duration. This study performed independent lifetime 502 ponent to the W matrix. Therefore, our study of the cathode experiments and comparisons for W-La cathodes, W-Ce cath- 503 needs to balance its ablation performance with the optimisa-448 odes, and pure W cathodes [43]. 449

The mass loss of the three experimental cathode materi- 505 als under the conditions of 3 h experimental duration with the 506 further analysed by energy spectrum as shown in Figure 6. plasma density up to  $1.0 \times 10^{19}$  m<sup>-3</sup> measured by Langmuir 507 Figures 6 a to 6 d show the atomic energy spectra and eleprobe and the temperature up to 1300°C measured by the bot- 508 mental analysis results of the W-La cathode before the extom plate temperature probe is shown in Figure 4b. Based 500 periment, in which the dark-coloured part is the area where 455 on parameters such as gas flow rate, input current, magnetic 510 the La<sub>2</sub>O<sub>3</sub> particles are located, which is highly overlapped 456 field strength, and plasma density, the service life of the W- 511 with the results of the point-scan elemental analysis in this 457 La cathode is estimated at approximately 3000 hours, and for 512 area, and the results of the face-scan elemental analysis in-

459 tributed to decreased thermal resistance of the cathode matrix 460 as rare earth element content increases, leading to higher mass loss under extreme operating conditions and consequently reduced service life.

Thus, in optimizing the performance of cathodes, it is expensive process, and even more expensive are tests of life 464 essential to ensure both excellent electron emission perfor-465 mance and high ablation resistance.

## Microstructural characterization of rare earth tungsten alloy cathodes

The melting, sputtering, and eventual deposition on the sur-469 face of the cathode material due to the W matrix are critical 471 terial. Preliminary analysis of the deposits shows no new el-472 ements, consisting solely of tungsten oxides, which exhibit valence changes at high temperatures [44].

The rare earth tungsten alloy cathode we prepared was 475 severely ablated after a long ignition experiment. The cathode 476 surface was characterised and analyzed by SEM before and after the experiment, as shown in Figure 5. Among them, Fig-478 ures 5 a to 5 c show the morphological characterisation analysis of W-La15 cathode before ignition experiment, the La<sub>2</sub>O<sub>3</sub> 480 particles are uniformly distributed on the smooth W substrate, and the elemental content distribution is shown in Figures 6 a 482 to 6 d. This indicates that the low fugitive work components 485 formly dispersed, which provides a favourable condition for 486 the improvement of the electron emission performance of the 487 cathode. Figures 5 c to 5 e show the morphological characteri-488 sation of the cathode after the ignition experiments, and it can 489 be clearly seen that the cathode tip hole shrinks severely due 490 to the prolonged ignition and plasma sputtering during the working process, which also leads to a more difficult cathode 492 ignition arc initiation under the same ignition power and pro-495 needs to be increased in order to break through the cathode Therefore, as shown in Figure 4c to 4e, we used our custom 496 and complete the ignition, which also causes greater damage 501 the cathode through the addition of a low fugitive work comtion of the ignition conditions.

The W-La cathodes before and after the experiment were 458 the W-Ce cathode, about 1100 hours. This reduction is at- 513 dicate that the main material of the whole cathode material

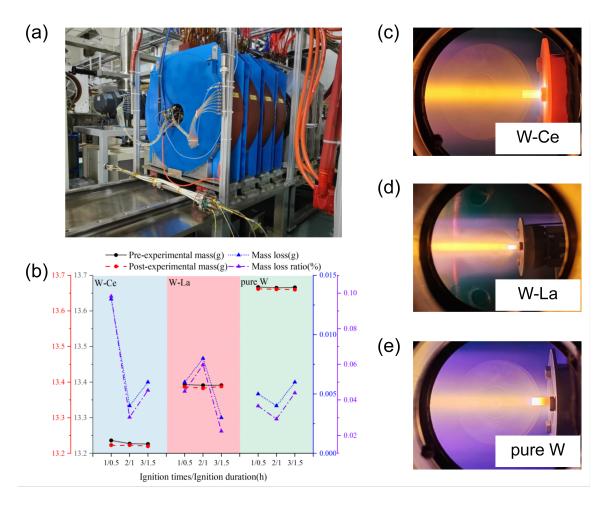


Fig. 4. Ablation life prediction experiment by cascade arc plasma generation source system.(a) Cascade arc plasma generation source system(Institute of Plasma Physics, Chinese Academy of Sciences, China); (b) Test cathodes mass loss data chart; (c) W-Ce cathode test; (d) W-La cathode test; (e) Pure W cathode test.

514 is tungsten, and the La<sub>2</sub>O<sub>3</sub> particles are uniformly dispersed 536 Moreover, the density, size, and evenness of such pores on 517 the W-La cathode after the experiment. The surface-scanning 539 material mixing methods. 518 elemental analysis results show that the content of W is as 519 high as 65.7%, followed by O element, and the content of 520 rare earth La is as low as 1.6%, which indicates that the La is 540 521 doped into W matrix in the form of W-La<sub>2</sub>O<sub>3</sub>, which is con-522 sumed in the process of discharging to complete the electron emission. Analysis of the particulate matter on the cathode surface in Figure 6 g shows that it is an oxide of W. The melting and recrystallisation of the W matrix at high temperatures changes the valence state to form different W oxides. 526

ide W-La<sub>2</sub>O<sub>3</sub>, due to its low work function, allows electrons 547 research on cathode needs to consider the doping elements near the Fermi level of La to overcome the surface barrier and 548 affecting the electron escape work of cathode materials, thus emit from the material under high voltage between the cath- 549 affecting the electron emission performance of cathode, in adode and anode. It results in the appearance of pores, which 550 dition, the ratio of doping elements and W matrix affects the 532 can be distributed on the surface of emitters in a very uneven 551 melting point of the overall alloy, that is, it affects the sputmanner, and it also demonstrates that the poor work function 552 tering resistance of cathode, etc.; in addition, the preparation 534 given by rare earth elements has been wildly added in order 553 method of cathode also greatly affects its performance, and 595 to improve the total performance and efficiency of emitters. 554 the optimization of the cathode preparation process enables

515 on the surface of the W substrate. Figures 6 e to 6 h show 537 the surface depend on many factors, including the size of the 516 the atomic energy spectra and elemental analysis results of 538 second phase particle, the doping mass fraction, and the raw

# V. RESULTS AND DISCUSSION

The cathode is one of the crucial components of various 542 electronic vacuum devices and space propulsion systems, and 543 the electron emission performance and ablation resistance of cathode materials against high-energy particle impacts directly affect the overall efficiency, performance and life of the The characterization results show that the rare earth ox- 546 system and other key indicators [45]. Therefore, the future

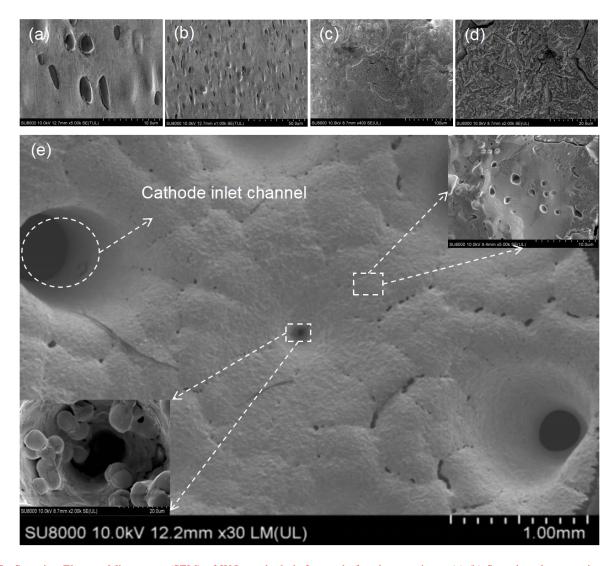


Fig. 5. Scanning Electron Microscopy (SEM) of W-La cathode before and after the experiment.(a)-(b) Scanning electron micrograph of W-La cathode before the experiment,  $5000 \, x$  and  $1000 \, x$ , respectively; (c)-(d) Scanning electron micrograph of W-La cathode after the experiment,  $400 \, x$  and  $2000 \, x$ , respectively; (e) Scanning electron micrograph of W-La cathode tip area after the experiment.

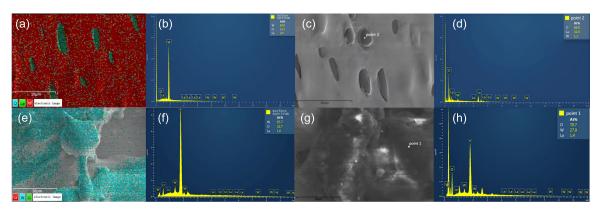


Fig. 6. Energy Dispersive Spectrum (EDS) of W-La cathode before and after the experiment.(a)-(d)Energy Dispersive Spectrum (EDS) of W-La cathode before the experiment, surface scanning and spot scanning, respectively; (e)-(h)Energy Dispersive Spectrum (EDS) of W-La cathode after the experiment, surface scanning and spot scanning, respectively.

555 the dopant phase to form a nano-sized in W matrix. disper- 556 sion, the better the dispersion, the better the improvement ef-

557 fect on the cathode emission performance.

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559 emission rare earth tungsten alloy cathode is investigated by 601 of Aeronautics and Astronautics, with a cumulative working both simulation and experimental verification. Atomic models 602 life of 8241 h [46]. Certainly, due to the more severe work-2.924 eV, and 2.908 eV for 0.5 ML0.5 ML of Ce, La, and 607 processing technology will have an impact on cathode life, 565 2.555 eV for 1.0 ML of Ce, La, and Y, correspondingly. These 609 to balance various factors, and the long-life high-performance duces the surface potential barrier of the cathode, which con- 611 focus. firms that doping of rare earth atoms is an effective method to enhance performances of electron emission for cathodes.

Some related studies have shown that the addition of 1% La<sub>2</sub>O<sub>3</sub> to W can improve its cutting performance and recrystallisation temperature, further optimising the WLa cathode sintering and processing. Of course, based on the Material Studio simulation result, in the paper, W-La, W-Ce, and W-Y cathodes with different treatment methods were prepared; a serial of ignition tests have been designed and performed; the results indicate that compared to pure tungsten cathodes, the operating condition for the rare earth tungsten alloy cathode operating is more stable with lower ignition voltage.

A cascade arc plasma source custom's development was <sup>583</sup> also employed to simulate the real working environment that tungsten alloy cathodes face. On the basis of all the same quantity of these operational parameters, namely, the gas flow rate, input current, magnetic field strength, and plasma density, the service lives for the considered rare earth tungsten alloy cathodes were estimated. In terms of cathode theoretical life prediction research, the more mature research programmes are for Hall thrusters and low-power ion thrusters, with less research on the cathodes of high-power, high-ratio 592 impulse magnetic plasma thrusters. The HET-80 Hall thruster 593 underwent life verification tests at the Shanghai Institute of 594 Space Propulsion (SISP) and the Beijing University of Aero-595 nautics and Astronautics (BUAA), respectively. The full life test was carried out in the VF-6 vacuum chamber of the 597 Shanghai Institute of Space Propulsion, with a cumulative 635 working time of 9240 h; the 1:1 working life test was car-

599 ried out in the DT-2.5 vacuum chamber of the Space Plasma In this paper, the optimization of high performance field 600 and Electric Propulsion Laboratory of the Beijing University are built with Material Studio for rare earth elements adsorbed 603 ing environment of high-power magnetic plasma propellers onto the W-O (top-site) surface; the relevant surface work 604 and the higher working temperature, the cathode life will be functions are then calculated based on density functional the- 605 greatly reduced. Therefore, factors such as cathode emitter ory. The work function values calculated were 2.645 eV, 606 material selection, ablation mechanism research and cathode Y adsorbed on the W-O top site and 2.770 eV, 3.356 eV, and 608 and the future cathode life optimisation research also needs results reflect that doping of rare earth atoms effectively re- 610 cathode for magnetic plasma thruster will also be a research

> Optical microscopy, SEM, and EDS were conducted on 613 the cathodes pre-experiment and pre-experiment and post-614 experiment. The experimental results obtained show that the 615 role of the rare earth doped in the tungsten alloy cathodes is 616 to excite the outer valence electrons at high voltage and break 617 down the propellant to form plasma. In continuous electron 618 emission, the consumption of the rare earth atoms is grad-619 ual. However, with the increase of the doping ratio of rare 620 earth, to a certain extent high-temperature resistance of the cathode declines and serious ablation occurs. Thus, by chang-622 ing the ratio of doping atoms in the cathode, an optimization of the trade-off between excellent electron emission performance and good resistance against ablation can be achieved.

> Based on the above research on different rare earth tung-626 sten alloy cathodes, we believe that high-performance field emission rare earth tungsten alloy cathodes will have great ap-628 plication prospects in many aspects such as vacuum devices, space propulsion, etc., in which the excellent electron emis-630 sion performance of tungsten-lanthanum cathode can greatly 631 improve the emission efficiency of the cathode itself. The ab-632 lation mechanism of rare earth tungsten alloy cathode and the 633 development of longer-life cathode will be the next research 634 focus.

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